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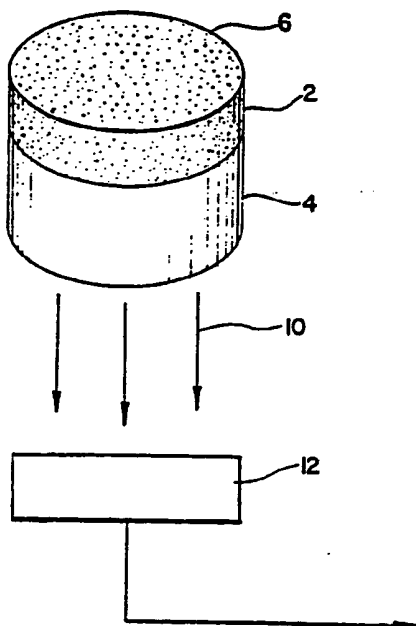
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(54) Title: **MEASUREMENT OF BACTERIAL CO₂ PRODUCTION IN AN ISOLATED FLUOROPHORE BY MONI-**
TORING AN ABSORBANCE REGULATED CHANGE OF FLUORESCENCE



(57) Abstract

The present invention is a multi-layer blood culture sensor which includes a pH sensitive absorbance dye encapsulated in or isolated by a first light transmissive gas permeable, proton impermeable matrix (2) and a pH insensitive fluorescence dye in a second matrix (4), the first and second matrices being spectrally coupled, and a method to use this sensor to detect microorganisms in a blood culture bottle.

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MEASUREMENT OF BACTERIAL CO₂ PRODUCTION IN AN ISOLATED FLUOROPHORE
BY MONITORING AN ABSORBANCE REGULATED CHANGE OF FLUORESCENCE

This is a continuation-in-part of U.S. Patent Application
Serial No. 7/609,278, filed November 5, 1990.

5 Field of the Invention

This invention relates to a method to detect the presence
or determine the concentration of microorganisms in a solution by
regulating light reaching or exiting from a fluorophore
encapsulated in or isolated by a chemically inert light
10 transparent matrix.

Background of the Invention

Microorganisms present in bodily fluid can be detected
using a culture bottle. Generally, a culture bottle is a flask
allowing positive cultures to be detected rapidly. The flask is
15 generally a transparent closed container filled with nutrient
that promotes the growth of the organism. In particular,
bacteria in blood can be detected in culture. U.S. Patent No.
4,772,558 (Hammann).

Many different qualitative and quantitative detection means
20 are used to monitor the growth of microorganisms in a culture
bottle. The microorganisms in a culture bottle have been
detected by use of external detectors such as a magnifying lens,
U.S. Patent No. 4,543,907 (Freudlich). Additionally, interal
detectors such as liquid level indicators can show bacterial
25 growth as a function of increased pressure in the vessel. Swaine
et al., EPA 124,193 (Swaine et al.). Additionally,
microorganisms can be detected by measuring changes in pH caused
by bacterial growth, Mariel, G.B. Patent No. 1,601,689.

Still another method to detect microorganisms involves the
30 use of a culture media that contains a compound which changes
color or appearance according to the growth of microorganisms.
The change in the media can be detected with a spectrophotometer.
There are many examples of reactions used in Microbiology that
rely on a color change. Bascomb, Enzyme Tests in Bacterial

Identification, 19 Meth. Microbio. 105 (1987). For example, a variety of organisms can be classified in large part by their pattern of fermentation, oxidation or assimilation of carbon sources. Fermentation of carbohydrates results in the production of acid which causes a decrease in pH. This drop in pH can be easily detected by including a pH indicator like bromothymol blue or phenol red. With both indicators, acid conditions representing the fermentation of a particular carbohydrate result in a yellow color (changing from blue-green for bromothymol blue or pink/red for phenol red). The same approach can be adopted for a variety of carbohydrates, ranging from monosaccharides like glucose to polysaccharides like inulin. In an analogous fashion, increasing pH can also be monitored. Assays for detecting the presence of decarboxylase and urease, and the ability to use malonate are based on an increase in pH, as indicated by a color change in the indicator. Turner, et al. U.S. Patent No. 4,945,060 discloses a device for detecting microorganisms. In this device changes in the indicator medium resulting from pH changes in CO₂ concentration in the medium are detected from outside the vessel.

Chemical and enzymatic reactions are used to detect or quantitate the presence of certain substances in microbiological or other assays. Many of these tests rely on the development or change of color or fluorescence to indicate the presence or quantity of the substance of interest.

Another approach to determine if an organism can degrade a particular substrate, is to use a reagent which is capable of reacting with one or more of the intermediates or final products. For example, the detection of the reduction of nitrate to nitrite. If nitrite is formed, then a pink to deep red color will result when sulfanilic acid and alpha-naphthylamine are added to the reaction mixture.

In contrast to the indirect detection of an enzymatic reaction illustrated by the nitrate/nitrite test, it is possible to use a synthetic analog of a natural substrate to directly

indicate the presence of an enzyme. For example, methylene blue can be reduced under certain conditions by the action of reductase, resulting in a shift from blue to colorless. In another test, the oxidase assay relies on the interaction of cytochrome oxidase with N, N, N', N'-tetramethyl-p-phenylenediamine producing a blue color.

Another example is the ability of microorganisms to degrade sulfur-containing amino acids as indicated by the production of H_2S . Typically, the organism is incubated with a high concentration of a sulfur-containing substrate (e.g. cysteine, cystine) in an acid environment. The production of H_2S is indicated by the formation of a black precipitate in the presence of ferric ammonium citrate.

Enzymes can usually act on more than one substrate. This allows for the use of synthetic enzyme substrates for the detection of enzyme activities. Synthetic substrates contain a metabolic moiety conjugated with a chromatic or fluorescent moiety. The conjugated molecule usually has a different absorption and/or emission spectrum from the unconjugated form. Moreover, the unconjugated chromatic or fluorescent moiety shows a considerably higher absorption or fluorescence coefficients than those of the conjugated molecule. This allows the measurement of small amounts of products of enzyme activities in the presence of the large amounts of conjugated substrate required for maximal enzyme activity. An example of a synthetic enzyme substrate is o-nitro-phenol- β -galactopyranoside used for the detection of activity of the enzyme β -galactosidase. The conjugated substrate is colorless. The β -galactosidase enzyme hydrolyzes the substrate to yield β -galactosidase and o-nitro-phenol. o-nitro-phenol absorbs strongly at 405nm, and its release can be measured by the increase in absorbance at that wavelength. Bascomb, Enzyme Tests in Bacterial Identification, Meth. Microbiol. 19, 105 (1987), reviewed the synthetic moieties used for enzyme substrates and the enzymatic activities measurable using this principle.

Presently, the monitoring of color or color end-product in chemical and microbial reactions is usually achieved in either of two ways; 1) the detection of color or color end-product can be achieved by visual observation and estimated qualitatively, or 2) the detection of color end-products or loss of color can be achieved by measuring the intensity of color instrumentally. Spectrophotometers that measure light absorbance are commonly used for this purpose. When measuring the concentration of a number of substances it is advantageous to use one instrument based on one principle of measurement, otherwise cost is increased.

Although the use of colorimetric reactions is widespread there are limitations, especially in the sensitivity of detection. In order to improve sensitivity and, in the case of identification of microorganisms, thereby to decrease the time required to obtain a result, fluorescence-based methods frequently are used. Unfortunately, it may not be possible to develop a fluorescent equivalent to every assay. Additionally, the fluorescent reagents themselves may be highly toxic and therefore difficult to commercialize.

In such cases one might need to measure activities of some enzymes fluorometrically, the others colorimetrically. However, most instruments are suited to measure either absorbance or fluorescence, and very few can be used to measure both.

The general principle of fluorescence quenching has been accepted as a way to detect or determine enzymatic or chemical reactions. For example, Fleminger et al. synthesized intramolecularly quenched fluorogenic substrates for the assay of bacterial aminopeptidase, P. Fleminger et al., Fluorogenic Substrates for Bacterial Aminopeptidase P and its Analogs Detected in Human Serum and Calf Lung, Eur. J. Biochem. 125, 609 (1982). In this case, the fluorescence of the aminobenzoyl group is quenched by the presence of a nitrophenylalanyl group. When the enzyme is present, the nitrophenylalanyl group is cleaved, with a concomitant increase in the sample's fluorescence. A

variety of enzymes have been assayed by this type of procedure, including hydrolytic enzymes, other amino- and carboxypeptidases and an endopeptidase. Yaron et al., Intramolecularly Quenched Fluorogenic Substrates for Hydrolytic Enzymes, Anal. Bioche. 95, 228 (1979); Carmel et al., Intramolecularly - Quenched Fluorescent Peptides as Fluorogenic Substrates of Leucine Aminopeptidase and Inhibitors of Clostridial Aminopeptidase, Eur. J. Biochem. 73, 617 (1977); Carmel et al., An Intramolecularly Quenched Fluorescent Tripeptide as a Fluorogenic Substrate of Angiotensin-I-Converting Enzyme and of Bacterial Dipeptidyl Carboxypeptidase, Eur. J. Biochem. 87, 265 (1978); Florentin et al., A Highly Sensitive Fluorometric Assay for "Enkephalinase", a Neutral Metalloendopeptidase that Releases Tyrosine-Glycine-Glycine from Enkephalins, Anal. Biochem 141, 62 (1984). In each of the previous approaches, a synthetic substrate containing a quenching group and a fluorescing group was generated in order to detect the activity of the enzyme.

An alternative to this approach would involve the synthesis of a resonance energy transfer pair of fluorescing groups on a substrate molecule. In this method, cleavage by the enzyme of one of the groups would result in a decrease in fluorescence, since the critical distance would be exceeded, eliminating the transfer of energy. However, the previously discussed approaches are limited to specifically designed substrates.

Still another approach involves the estimation of a chromophore by fluorescence measurement. See W. Blumberg et al., Hemoglobin Determined in Whole Blood "Front Face" Fluorometry, Clin. Hemo. 26, 409 (1980). Blumberg disclosed an assay based on attenuation of fluorescence of a dye, whose excitation wavelengths overlap with the absorption wavelengths of the chromophore.

Subsequently, M. Shaffer, U.S. Patent No. 4,495,293 (hereinafter Shaffer) filed a patent application disclosing a method to fluorometrically determine a ligand in an assay solution using conventional fluorometric techniques. In Shaffer

the intensity of the fluorescence emitted by the assay solution is related to the change in transmissive properties of the assay solution produced by the interaction of the ligand to be determined and a reagent system capable of producing change in the transmissive properties of the assay solution in the presence of the ligand. More particularly, Shaffer discloses a method to monitor absorbance using a fluorophore in solution with the chromophore. In this method the fluorophore may interact with the assay cocktail and produce changes in fluorescence intensity which are unrelated to the change being measured. The selection of the fluorophores is also restricted, in that pH dependent or environment sensitive fluorophores cannot be utilized. Additionally, when the fluorophore is in solution, less than accurate measure of absorbance may be obtained because light is absorbed exponentially through the chromophore sample.

Similarly, Beggs & Sand, EPA 91,837 disclosed a solution based method for determination of tryptophan-deaminase activity by measuring the reduction in fluorescence in the presence of a chromophore produced by the interaction between indole pyruvic acid and metal ions using a fluorophore "whose fluorescence is capable of being quenched by the indole pyruvate-metal ion complex, the ions of the fluorophore being present throughout the incubation period".

Also, Sands, U.S. Patent No. 4,798,788 discloses a process to detect a nitrate reducing microorganism by measuring reduction of fluorescence in solution by causing the diazotization of the fluorophore. In all these cases a specific fluorophore needs to be chosen for each test to ensure that it will fluoresce under the conditions of the test. e.g. only few fluorophores fluoresce at pH of less than 2.0.

Summary of the Invention

This invention provides a method to use a fluorophore enclosed or embedded in a chemically inert matrix which is transparent at the wavelengths of interest. The fluorophore,

positioned to intersect the transmission light path, indirectly monitors absorbance or changes in the absorbance of a chromophore encapsulated or isolated by a gas permeable polymeric matrix. The use of a fluorophore encapsulated in or isolated by a matrix allows for the sequential influence of reaction components on the intensity of light detected. This result can be achieved when the absorption spectrum of a chromophore overlaps the excitation and/or the emission spectrum of a fluorophore, thereby allowing the change in fluorescence to be related to the intensity of color in the reaction and consequently related to the quantity of the substance of interest. It should be noted that the spectrum is not limited to visible light.

More particularly, this invention relates to a multi-layer body fluid culture sensor comprised of a pH sensitive absorbance based dye spectrally coupled to a pH insensitive, or pH sensitive dye that is highly buffered, fluorescence based dye. The pH sensitive absorbance based dye is encapsulated or isolated in a polymeric layer that is permeable to CO_2 and water, but impermeable to protons. The pH insensitive fluorophore is encapsulated or isolated in the second polymeric layer that may or may not be permeable to CO_2 and water. This type of sensor may be used to detect or determine the concentration of microorganisms in bodily fluid. The spectral criterion required to make this determination are such that the absorption spectrum of the chromophore must overlap the excitation and/or emission spectrum of the fluorophore, thereby allowing the change in fluorescence to be related to the change in the reaction and consequently related to the presence or quantity of the substance of interest.

Further, this sensor is used to monitor microbial infections grown in a fluid culture bottle. In particular, this sensor can be used to monitor bacterial growth. As bacteria grow they generate CO_2 . The CO_2 generated by the bacteria diffuses into the polymeric layer that is in direct contact with a hydrated pH sensitive absorbance based dye. The CO_2 reacts with the aqueous environment to form carbonic acid (H_2CO_3), which

lowers the pH of the absorbance dye environment. This results in a concomitant change in the pH sensitive spectrum of the dye. Typically, as the absorbance of an absorbance based dye decreases more light reaches the fluorophore for excitation which results in a larger amount of emitted fluorescence.

In one embodiment the gas permeable, proton impermeable polymeric matrix is silicone. Additionally, in one embodiment of this invention a detector, such as a photomultiplier tube, is placed under the blood culture bottle to detect fluorescent emission.

Brief Description of the Figures

FIG.1 shows a schematic diagram of a multi-layer blood culture sensor.

FIG.2 shows a blood culture growth curve detected by a xylenol blue-rhodamine 101 sensor.

FIG.3 shows a blood culture growth curve detected by xylenol blue in silicone-rhodamine B in acrylic sensor.

FIG.4 shows a blood culture growth curve for a xylenol blue in silicone-6213 acrylic sensor.

FIG.5 shows a blood culture growth curve for a bromothymol blue in silicone-rhodamine 101 in silicone sensor.

Detailed Description - Best Mode

In this approach, fluorescence from a fluorophore embedded in an inert light-transparent matrix, is modulated by a pH sensitive absorbance dye embedded in a polymeric gas permeable, but proton impermeable matrix. The assay is carried out in a blood culture bottle.

In a fluorometric based colorimetric assay the fluorescence intensity is regulated by changes in absorbance of an interfering chromophore. As a pH change occurs the chromophoric material alters the amount of light reaching the fluorophore and/or the amount of emitted light reaching the detector. Spectrally compatible fluorescent and colorimetric indicators are selected so that as the pH changes due to the production of CO_2 by microorganisms present in the sample, the colorimetric indicators regulate the amount of light reaching the fluorophore and/or

photodetector and, thus cause a change in the excitation and/or emission of the fluorescent dye. This change is detected with a fluorescent reader and can be correlated with the presence or concentration of microorganism in the sample.

5 A bodily fluid culture sensor is comprised of a pH sensitive absorbance based dye in or isolated by a polymeric gas permeable, but proton impermeable matrix, and a fluorescent dye in a second polymeric matrix.

10 Spectrally compatible fluorescent and colorimetric indicators are selected so that when an organism is present in a sample, the colorimetric indicator will regulate the amount of light reaching the fluorophore thereby causing a change in the emission intensity from the fluorescence dye reaching the photodetector. The change, indicating the presence of bacteria,
15 is detected with a fluorometric reader.

 More particularly, spectrally compatible fluorescence and absorbance dyes are selected so that as carbonic acid is produced (CO_2 and H_2O), the absorbance of the dye will change thereby regulating the amount of light reaching the fluorophore and/or
20 photodetector, thus producing a change in the measured fluorescence. This change is detected with a fluorescence reader. Spectrally compatible dyes are rhodamine b and xylenol blue. Additionally, promothymol blue and rhodamine 101 are also spectrally compatible.

25 Thus, in practice a culture bottle containing the appropriate growth media can be inoculated with E. coli. As the organism grows, it produces CO_2 gas. The silicone is permeable to the CO_2 . The CO_2 diffuses to the absorbance layer and reacts with water to produce carbonic acid (H_2CO_3). The carbonic acid
30 causes a drop in the pH in the absorbance dye environment resulting in a change in measured absorbance. For example, as the pH drops in an absorbance layer containing the dye xylenol blue, the absorbance of xylenol blue decreases, thereby allowing more light to reach the fluorophore to excite it and thus
35 increase the amount of fluorescence emitted at 590nm. A positive culture using xylenol blue as the absorbance dye is detected by

increases the amount of fluorescence emitted at 590nm. A positive culture using xylenol blue as the absorbance dye is detected by a measured increase in fluorescence as the xylenol blue decreases in absorbance.

5 The pH sensitive absorbance based dye is encapsulated in or isolated by a polymeric matrix that is gas permeable, but proton impermeable. The polymeric matrix must be optically transparent in the visible region, permeable to gas, autoclavable, stable for at least six months, and proton impermeable. In particular,
10 silicone may function as the polymeric matrix used to encapsulate or isolate the absorbance based dye. Silicones found to meet these criteria were Dow, Rhone Poulenc, G.E. and Wacker.

 Similarly, the fluorescence based dyes can also be encapsulated in a polymeric matrix. The polymeric matrix used
15 for the fluorophore does not have to meet all of the above requirements listed for the matrix used to encapsulate or isolate the absorbance dye. The similar features that it must possess are that it must be optically transparent in the visible region, autoclavable and stable for at least six months.

20 The polymeric matrix containing or isolating the absorbance based dye must be coupled to the polymeric matrix containing the fluorescent dye. It should be noted that the polymeric matrices must be in close proximity so that light that has been regulated by the absorbance layer will have an effect on the emission
25 intensity of the fluorophore as received by the photodetector. This can be accomplished by applying the same polymeric material to one side of each polymeric matrix and curing these matrices. Once the matrices containing the dyes have been adhered together they must be rehydrated. The clarity of the sensor upon
30 rehydration is also a factor in matrix selection.

 In the present invention, a bodily fluid culture sensor, FIG.1, is comprised of a pH sensitive absorbance based dye encapsulated in or isolated by a polymeric gas permeable, but proton impermeable matrix 4 and a fluorescent dye in a second
35 polymeric matrix 2. Reflective surface 6 can be included to

facilitate the transmission of light to the detecting element 12. In FIG.1 interrogation light enters the sensor and is regulated by pH sensitive matrix 2 which in turn causes a change in the fluorescence emission 10 of the fluorophore in matrix 4. This sensor offers the advantage of maximal surface area.

In an alternative embodiment, an acrylic encapsulated fluorophore or silicone embedded fluorescence material is adhered to an absorbance dye isolating polymeric layer, to make a two layer sensor.

In another embodiment, both the fluorescence and absorbance embedded material are poured into blood culture bottles. In this embodiment the fluorophore embedded silicone material is poured on top of absorbance embedded silicone.

The optical interrogation system comprises a visible output, 400-700nm, light source focused onto one end of a bifurcated fiber optic cable. The common end is positioned close to the sensor, while the other end is positioned close to a photodetector, typically a photomultiplier tube. Appropriate excitation and emission filters are used to select wavelengths of choice for each dye. A beam splitter is used to divert a portion of the excitation light to a second photodetector and acts as a reference. A photodetector converts light to a current source which is converted to a voltage using an operational amplifier. A 12 bit analog to digital conversion offers sufficient dynamic range to read the voltage. A computer program is then used to read, plot and store data.

A measurement is taken by first reading reference light intensity. Next the reading from the sensor disk is measured. The data is plotted by taking the ratio of reference, excitation light, to sample. In particular, as CO_2 levels increase in the blood culture bottle, the absorbance of the absorbance dye changes, thereby changing the amount of light reaching the fluorescence layer and/or photodetector. This causes a change in emitted fluorescence that is detected.

The following examples serve to illustrate the method of the present invention. The concentration of reagents and other variable parameters are only shown to exemplify the methods of the present invention and are not to be considered limitation thereof.

Example 1 Xylenol Blue - Rhodamine 101 Sensor

Wacker silicone elastomer 3601 part A is thoroughly mixed with Wacker 3601 catalyst part B in a 9:1 ratio, as recommended by the manufacturer. Next 5% w/w of a 50mM xylenol blue, dissolved in 5mM borate buffer pH 11 containing 1% Tween 80, is added to the silicone and homogenized to ensure a uniform distribution of the dye. The absorbance layer mixture is then poured into an aluminum square mold to a thickness of 30/1000 of an inch and cured at 55°C for 2 hours.

Wacker silicone is prepared, as described above. Next 2% w/w of 7.5mM Rhodamine 101, in 50mM Tris-HCl buffer pH 8.5 in 95% ethylene glycol, is added to the silicone. The mixture is poured over the previously cured xylenol blue layer in the mold, described above, and cured at 55°C overnight. This cured, dehydrated, double layer sensor consists of two distinct layers, each 30/1000 of an inch thick. Disks may now be punched out of the mold and adhered onto the base of bottles using more silicone, ensuring that the absorbance layer is face down. Finally, the bottles are cured at 55°C for 15 minutes, rehydrated with normal saline and autoclaved on the wet cycle for 17 minutes. Saline is replaced with growth media and inoculated with E. coli by injecting a suspension with a sterile needle through the septum. The blood culture bottle is placed in the instrument and fluorescence emission is measured.

As the concentration of CO₂ increases in the blood culture bottle, the pH sensitive absorbance dye, Xylenol blue, the absorbance of the dye decreases, thus allowing more light to reach the fluorophore, Rhodamine 101, to thus increase the amount of fluorescence emitted at 690nm. This increase in fluorescence intensity v. time is shown in the blood culture growth curve at FIG.2.

Example 2 Xylenol Blue in Silicon /Rhodamine B in Acrylic

Rhone Poulenc silicone elastomer 141 part A is thoroughly mixed with Rhone Poulenc 141 catalyst part B in a 10:1 ratio, as recommended by the manufacturer. Next 1% w/w of a 100mM xylenol blue solution pH #11, dissolved in 10 mM borate buffer containing 1% Tween 80, is added to the silicone and mixed thoroughly with a tongue blade to ensure uniform distribution of the dye. The absorbance layer mixture is then poured into an aluminum square mold to a thickness of 30/1000 of an inch. The mold is allowed to sit out on the countertop at room temperature for about one hour or until the bubbles have disappeared, at which time the mold is placed in the incubator to cure at 55°C for two hours.

Rhone-Poulenc silicone is prepared, as described above. Next, a 40/1,000" thick acrylic disc (Glasflex, Inc.), approximately 1 cm in diameter, containing 0.2 grams/lb of rhodamine B (Sigma) is glued onto the above absorbance layer using the Rhone-Poulenc silicone at the 10/1 ratio as glue. The double layer sensor is then placed back in the 55°C incubator for two hours to allow for adherence of the two layers. Following the curing, the double layer sensor is punched out with a cork borer, and glued onto the base of a Wheaton bottle, ensuring that the absorbance layer is face down, using the Rhone Poulenc silicone as mentioned above. The bottle is placed in the 55°C incubator to cure for at least two hours. The bottle is then rehydrated overnite and tested the following day as described in Example 1.

As the concentration of CO₂ increases in the blood culture bottle, the absorbance of the pH sensitive absorbance based dye xylenol blue decreases, thus allowing more light to reach the fluorophore (rhodamine B) doped acrylic, to thus increase the amount of fluorescence emitted at 590nm. This increase in fluorescence intensity v. time is shown in the blood culture growth curve in FIG.3.

Example 3 Xylenol Blue in Silicone/6213 Red Standard Acrylic

Wacker silicone elastomer 3601 part A is thoroughly mixed with Wacker 3601 catalyst part B in a 9:1 ratio, as recommended

with Wacker 3601 catalyst part B in a 9:1 ratio, as recommended by the manufacturer. Next 5% w/w of a 50mM xylenol blue, dissolved in 5mM borate buffer pH 11 containing 1% Tween 80, is added to the silicone and homogenized to ensure a uniform distribution of the dye. The absorbance layer mixture is then poured into an aluminum square mold to a thickness of 30/1000 of an inch and cured at 55°C for two hours.

Next, a 40/1,000" thick acrylic disc (Glasflex, Inc.), approximately 1 cm in diameter, referred to as No. 6213 Red (Glasflex Standard Product) is glued onto the above absorbance layer using the Wacker silicone at the 9/1 ratio as glue. The double layer sensor is then placed back in the 55°C incubator for two hours to allow for adherence of the two layers. Following the curing, the double layer sensor is punched out with a cork borer, and glued onto the base of a Wheaton bottle, ensuring that the absorbance layer is face down, using the Rhone Poulenc silicone as mentioned above. The bottle is placed in the 55°C incubator to cure for at least two hours. The bottle is then rehydrated overnight and tested the following day as described in Example 1.

As the concentration of CO₂ increases in the blood culture bottle, the absorbance of the pH sensitive absorbance based dye xylenol blue decreases, thus allowing more light to reach the fluorophore (rhodamine b) doped acrylic, to thus increase the amount of fluorescence emitted at 590nm. This increase in fluorescence intensity v. time is shown in the blood culture growth curve in FIG.4.

Example 4 Bromothymol Blue in Silicone/Rhodamine 101 in Silicone

Wacker silicone elastomer 3601 part A is thoroughly mixed with Wacker 3601 catalyst part B in a 9:1 ratio, as recommended by the manufacturer. Next 5% w/w of 50mM bromothymol blue, dissolved in 5mM tris buffer pH 12 in ethylene glycol, is added to the silicone and homogenized to ensure a uniform distribution of the dye. The absorbance layer mixture is then poured into an

aluminum squar mold to a thickness of 30/1000 of an inch and cured at 55°C for two hours.

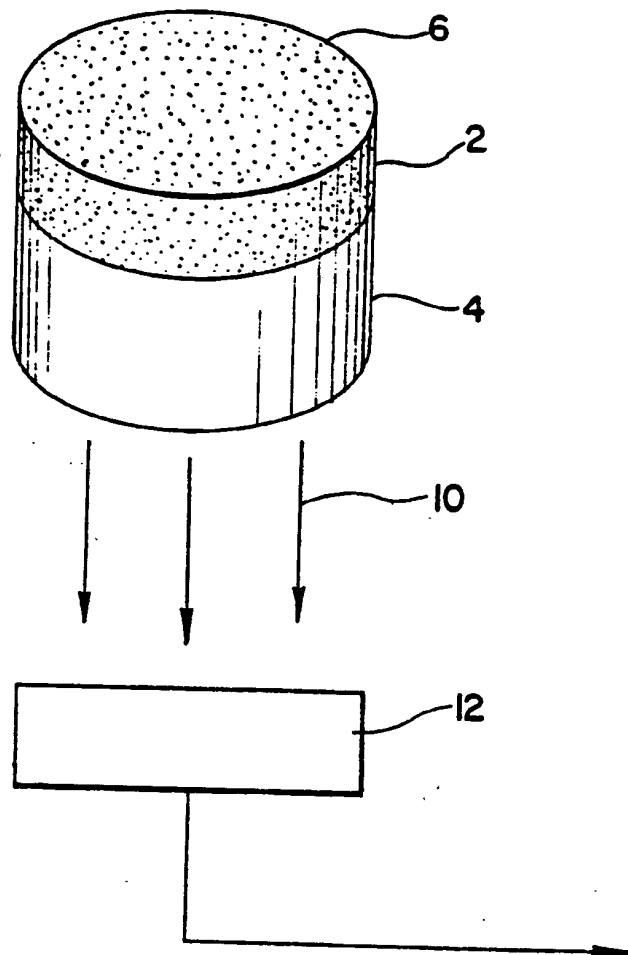
Wacker silicone is prepared, as described above. Next 2% w/w of 7.5mM Rhodamine 101, in 50mM Tris-HCl buffer pH 8.5 in 95% ethylene glycol, is added to the silicone. The mixture is poured over the previously cured xylenol blue layer in the mold, described above to isolate the absorbance layer. This sensor is then cured at 55°C overnight. This cured, dehydrated, double layer sensor consists of two distinct layers, each 30/1000 of an inch thick. Disks may now be punched out of the mold and adhered onto the base of bottles using more silicone, ensuring that the absorbance layer is face down. Finally, the bottles are cured at 55°C for 15 minutes, rehydrated with normal saline and autoclaved on the wet cycle for 17 minutes. Saline is replaced with growth media and inoculated with E. coli by injecting a suspension with a sterile needle through the septum. The blood culture bottle is placed in the instrument and fluorescence emission is measured. The increase in fluorescence intensity v. time is shown in blood culture growth curve in FIG.5.

Although this invention has been described with respect to specific embodiments, the details thereof are not to be construed as limitations, for it will be apparent that various equivalents, changes and modifications may be resorted to without departing from the spirit and scope thereof and it is understood that such equivalent embodiments are intended to be included herein.

WHAT IS CLAIMED IS:

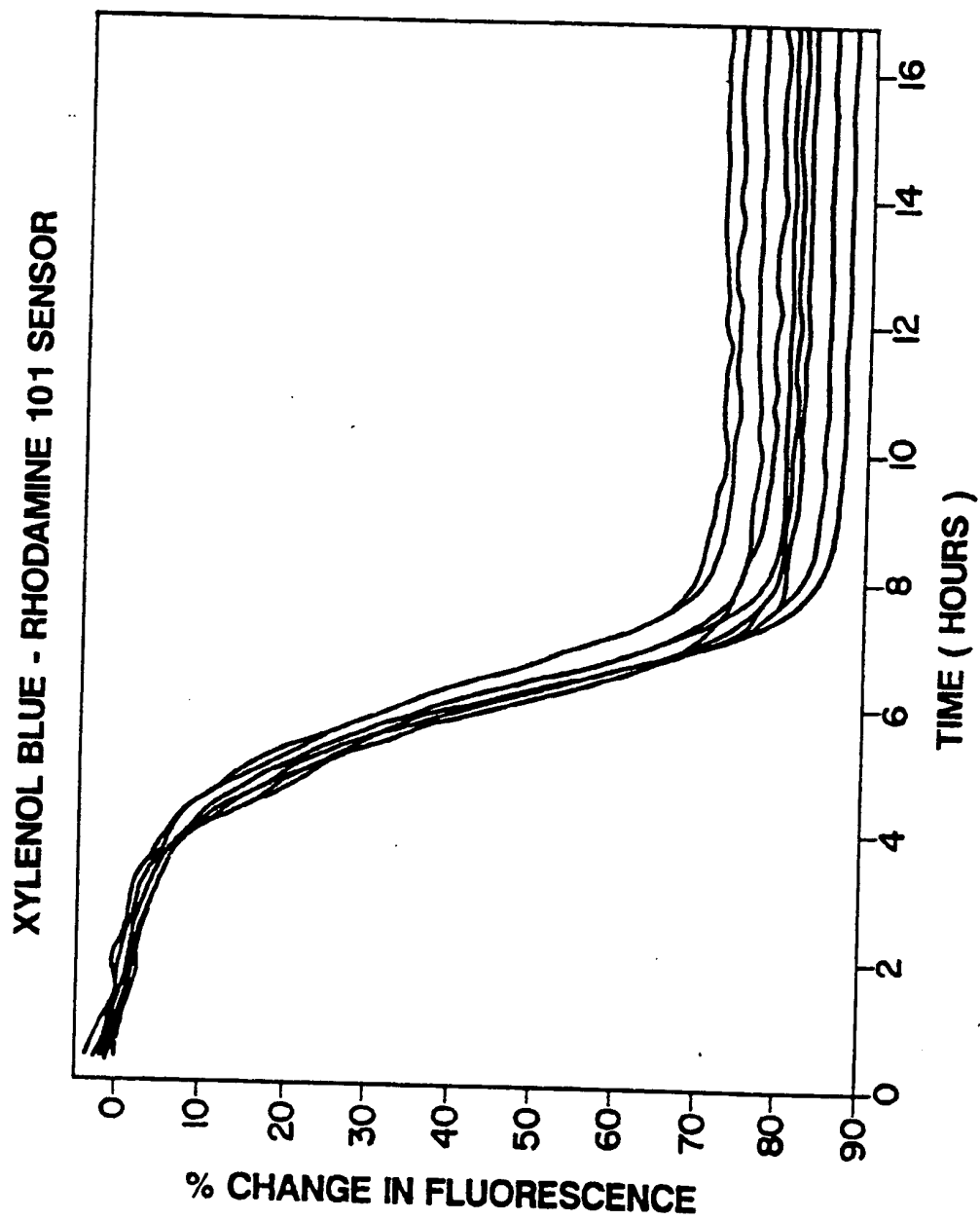
1. A multi-layer blood culture sensor which comprises:
 - a. a pH sensitive absorbance based dye encapsulated in a first light transmissive, gas permeable, proton impermeable matrix; and
 - 5 b. a pH insensitive fluorescence dye in an inert light-transparent matrix, said first and second matrices being spectrally coupled.
2. The blood culture sensor of Claim 1 wherein said pH sensitive absorbance based dye is selected from the class
10 consisting of xylenol blue or bromothymol blue.
3. The blood culture sensor of Claim 1 wherein said fluorescence dye is selected from the class consisting of Rhodamine B or Rhodamine 101.
4. The blood culture sensor of Claim 1 wherein said matrices
15 are selected from the class consisting of silicone or acrylic.
5. A multi-layer blood culture sensor which comprises:
 - a. a pH sensitive absorbance based dye isolated by a first light transmissive, gas permeable, proton
20 impermeable matrix; and
 - b. a pH insensitive fluorescence dye in an inert light-transparent matrix, said first and second matrices being spectrally coupled.
6. The blood culture sensor of Claim 5 wherein said pH
25 sensitive absorbance based dye is selected from the class consisting of xylenol blue or bromothymol blue.
7. The blood culture sensor of Claim 5 wherein said fluorescence dye is selected from the class consisting of Rhodamine B or Rhodamine 101.
8. The blood culture sensor of Claim 5 wherein said matrices
30 are selected from the class consisting of silicone or acrylic.

- 5
9. A method to detect or determine the concentration of microorganism in a bodily fluid culture bottle comprising:
- a. adding bodily fluid to a culture bottle containing the sensor of Claim 1 or Claim 5;
 - b. detecting fluorescent emission; and
 - c. correlating the change in fluorescence intensity with the presence or concentration of said microorganism.
- 10
10. The method of Claim 9 wherein said pH sensitive absorbance based dye is selected from the class consisting of xylenol blue or bromothymol blue.
- 15
11. The method of Claim 9 wherein said fluorescent dye is selected from the class consisting of Rhodamine B or Rhodamine 6G.
12. The method of Claim 10 wherein said matrices are selected from the class consisting of silicone or acrylic.
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- 35

Fig. 1

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Fig. 2



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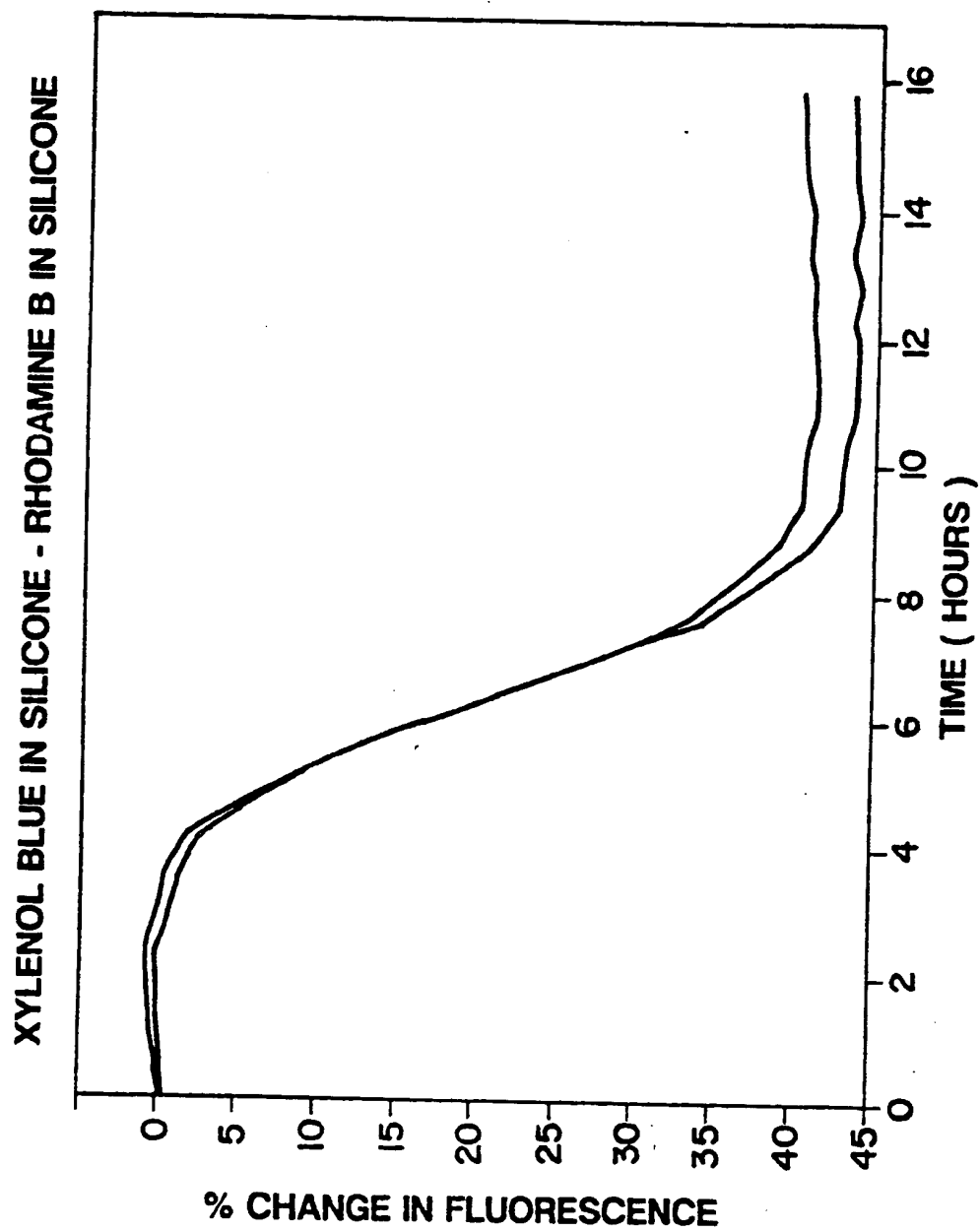
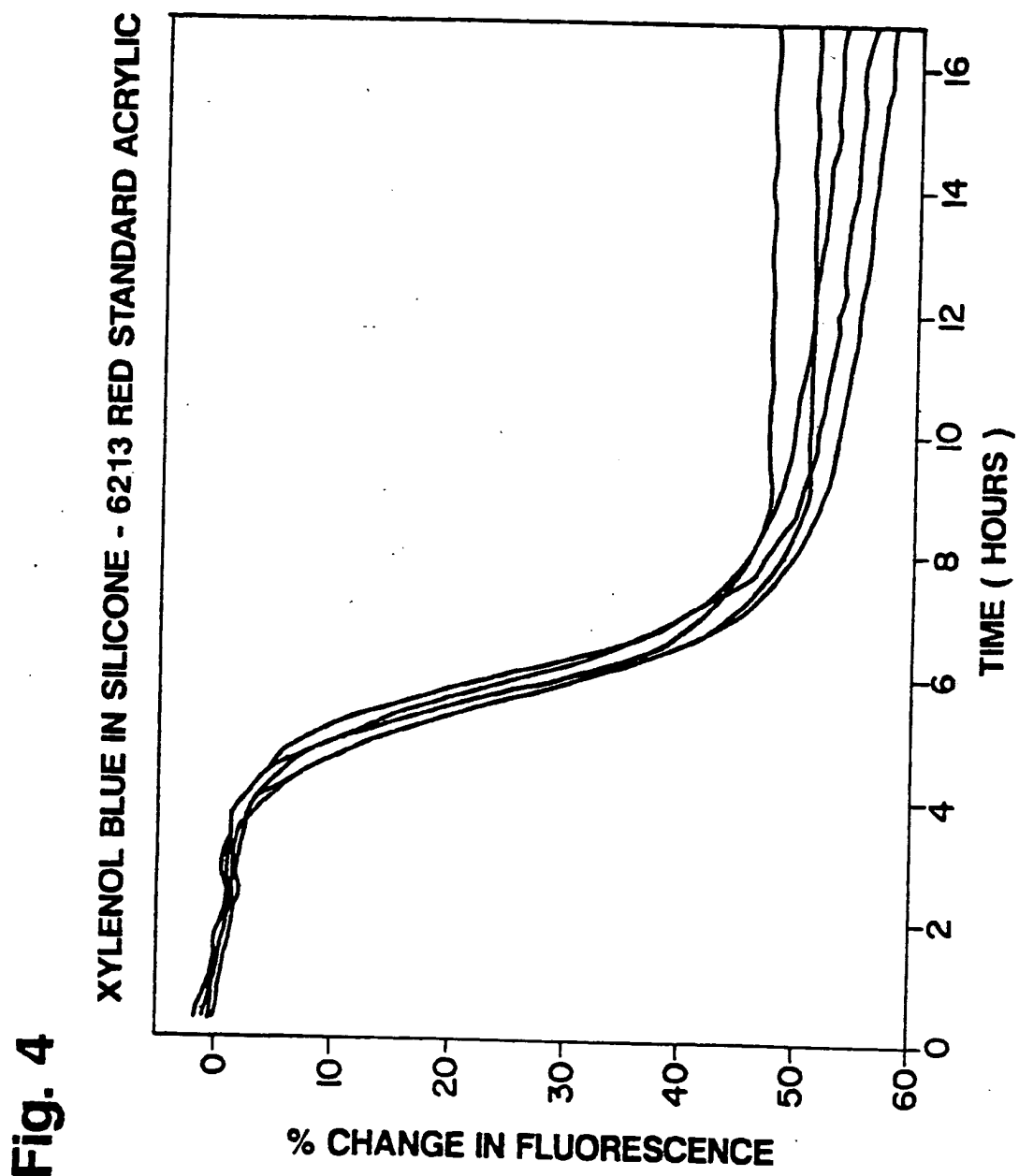
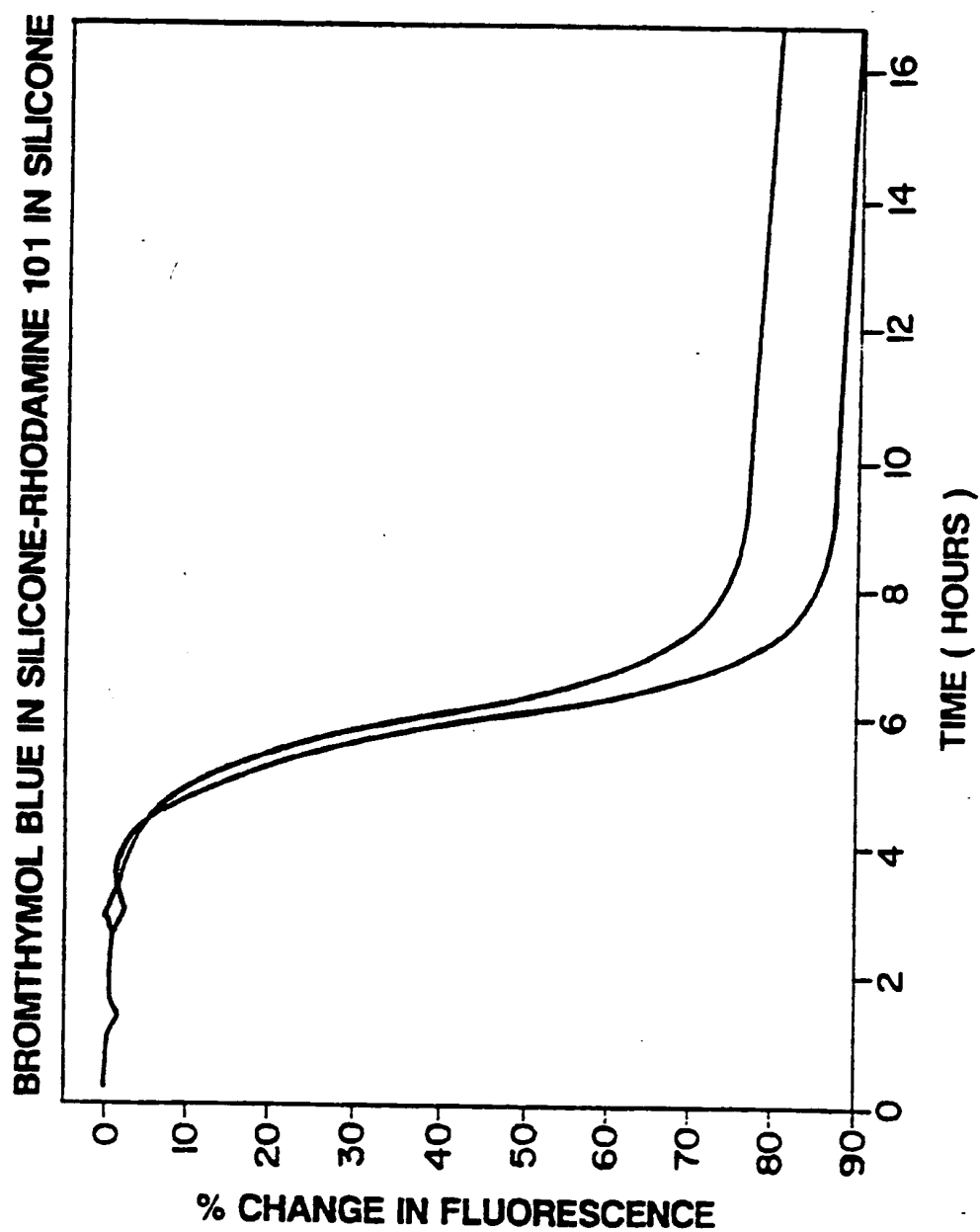


Fig. 3

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**Fig. 5**

INTERNATIONAL SEARCH REPORT

International Application No. PCT/US91/09716

I. CLASSIFICATION OF SUBJECT MATTER (if several classification symbols apply, indicate all) ³		
According to International Patent Classification (IPC) or to both National Classification and IPC IPC (5): G01N 21/64, 21/76; C12Q 1/06 US CL : 422/82.07, 82.08; 435/39; 436/172		
II. FIELDS SEARCHED		
Minimum Documentation Searched ⁴		
Classification System	Classification Symbols	
U.S.	422/52, 82.05, 82.07, 82.08; 435/39, 34, 808, 291; 436/172, 167, 169, 170, 800	
Documentation Searched other than Minimum Documentation to the extent that such Documents are included in the Fields Searched ⁶		
III. DOCUMENTS CONSIDERED TO BE RELEVANT ¹⁴		
Category ⁵	Citation of Document, ¹⁶ with indication, where appropriate, of the relevant passages ¹⁷	Relevant to Claim No. ¹⁸
Y	US, A, 4,851,195 (Matthews et al.) 25 July 1989. See detailed description of the invention.	1,5
Y	US, A, 4,231,754 (Vogelhut) 04 November 1980. See claims 1 and 4	1,5
Y	US, A, 4,945,060, (TURNER et al) 31 July 1990. See claims 1 and 2.	2,6,9,10
Y	US, A, 4,803,049 (Hirschfeld et al.) 07 February 1989. See claim 1.	3,7,11
Y	US, A, 4,867,919 (Yasfuso et al.) 19 September 1989. See brief description of the invention.	4,8,12
<div style="display: flex; justify-content: space-between;"> <div style="width: 45%;"> <p>¹⁵ Special categories of cited documents:</p> <p>"A" document defining the general state of the art which is not considered to be of particular relevance</p> <p>"E" earlier document but published on or after the international filing date</p> <p>"L" document which may throw doubts on priority claim(s) or which is cited to establish the publication date of another citation or other special reason (as specified)</p> <p>"O" document referring to an oral disclosure, use, exhibition or other means</p> <p>"P" document published prior to the international filing date but later than the priority date claimed</p> </div> <div style="width: 50%;"> <p>"T" later document published after the international filing date or priority date and not in conflict with the application but cited to understand the principle or theory underlying the invention</p> <p>"X" document of particular relevance; the claimed invention cannot be considered novel or cannot be considered to involve an inventive step</p> <p>"Y" document of particular relevance; the claimed invention cannot be considered to involve an inventive step when the document is combined with one or more other such documents, such combination being obvious to a person skilled in the art</p> <p>"&" document member of the same patent family</p> </div> </div>		
IV. CERTIFICATION		
Date of the Actual Completion of the International Search ²		Date of Mailing of this International Search Report ²
24 MARCH 1992		31 MAR 1992
International Searching Authority ¹		Signature of Authorized Officer ²⁰
ISA/US		Maureen Wallenhorst ebw